Closure of the Yeast Mitochondria Unspecific Channel (YMUC) Unmasks a Mg^{2+} and Quinine Sensitive K^+ Uptake Pathway in Saccharomyces cerevisiae

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The K^+ uptake pathways in yeast mitochondria are still undefined. Nonetheless, the K^+ -mediated mitochondrial swelling observed in the absence of phosphate (PO₄) and in the presence of a respiratory substrate has led to propose that large K^+ movements occur in yeast mitochondria. Thus, the uptake of K^+ by isolated yeast mitochondria was evaluated. Two parallel experiments were conducted to evaluate K^+ transport; these were mitochondrial swelling and the uptake of the radioactive K^+ analog 86 Rb $^+$. The opening of the yeast mitochondrial unspecific channel (YMUC) was regulated by different PO₄ concentrations. The high protein concentrations used to measure 86 Rb $^+$ uptake resulted in a slight stabilization of the transmembrane potential at 0.4 mM PO₄ but not at 0 or 4 mM PO₄. At 4 mM PO₄ swelling was inhibited while, in contrast, 86 Rb $^+$ uptake was still observed. The results suggest that an energy-dependent K^+ uptake mechanism was unmasked when the YMUC was closed. To further analyze the properties of this K^+ uptake system, the Mg^{2+} and quinine sensitivity of both swelling and 86 Rb $^+$ uptake were evaluated. Under the conditions where the unspecific pore was closed, K^+ transport sensitivity to Mg^{2+} and quinine increased. In addition, when Zn^{2+} was added as an antiport inhibitor, uptake of 86 Rb $^+$ increased. It is suggested that in yeast mitochondria, the K^+ concentration is highly regulated by the equilibrium of uptake and exit of this cation through two specific transporters.

KEY WORDS: Permeability transition; K⁺ transport; K⁺ channels; yeast mitochondria; YMUC.

INTRODUCTION

In the cell, K^+ is the most abundant cation, both in the cytoplasm and within the mitochondrial matrix, where its concentration has been estimated at 140 mM (for a review, see Brierley *et al.*, 1994). K^+ is perhaps the main mitochondrial osmoregulator, and the volume of the matrix is modified in response to the K^+ movements across the mitochondrial inner membrane. In mammalian mitochondria, the uptake of K^+ was first proposed to occur through a 53 kDa uniporter (Diwan *et al.*, 1988). More recently, a mitochondrial K_{ATP} channel was identified which is similar to the plasma membrane uptake system (Inoue *et al.*, 1991); the K^+ uniport activity seems to have little

dependence on pH although it exhibits high sensitivity to adenine nucleotides (Beavis et al., 1993). The mitochondrial and the plasma membrane KATP channels are both closed by ATP or glybenclamide (Garlid, 1996) and probably by ruthenium red (Kapus et al., 1990). The exit of K⁺ is catalyzed through an antiport with protons, as predicted by Mitchell (1961) and characterized by Brierley (1978). These two K⁺ transport systems control the concentration of K⁺ within isolated mitochondria (Garlid, 1980). In turn, mitochondrial K⁺ has large effects on mitochondrial volume, thus regulating diverse processes, such as oxidative phosphorylation and fatty acid catabolism (Halestrap, 1989; Nicholls et al., 1972). The release of K⁺ is activated when mitochondrial Mg²⁺ or nucleotides are depleted, or when the pH is increased (Jezek et al., 1990). It has been proposed that the recycling of K⁺ needs to be highly regulated in order to avoid the depletion of both the transmembrane potential and the pH gradient (Garlid, 1988).

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Mitochondria are regarded as perfect osmometers (Chappel and Crofts, 1966; Roucou *et al.*, 1995). The swelling method has been used to study many mitochondrial transport systems including the K⁺/H⁺ antiporter (Beavis *et al.*, 1993; Brierley *et al.*, 1994). Using this method, several features of the mammalian K⁺/H⁺ antiporter have been defined (Garlid, 1988): it is inhibited by Mg²⁺ (Jung and Brierley, 1986), by dicyclohexylcarbodiimide (DCCD) (Martin *et al.*, 1986) and by quinine (Brierley and Jung, 1988; Nakashima and Garlid, 1982). In addition, it has been demonstrated that quinine exhibits a partial inhibitory effect on the K⁺ uniporter (Diwan, 1986).

The mitochondrial swelling method also allowed the characterization of the K⁺/H⁺ antiporter in mitochondria from *Saccharomyces cerevisiae*, which is similar to the antiporter from mammalian mitochondria, except that in yeast inhibition by Mg²⁺ is not detected and transport is active at physiologic pH (Roucou *et al.*, 1995; Welihinda *et al.*, 1993). In addition it was demonstrated that the yeast K⁺ antiporter is sensitive to Zn²⁺ or Cu²⁺ (Manon and Guérin 1992, 1995).

The use of electrophysiological techniques has revealed the existence of high-conductance, unspecific ionic channels in the internal membrane of yeast mitochondria (Ballarin and Sorgato, 1995, 1996). These unspecific channels have been characterized physiologically and have been tentatively identified as the equivalent to the permeability transition pore (PTP) from mammalian mitochondria (Zoratti and Szabó, 1995). Both the PTP and the yeast mitochondrial unspecific channel (YMUC) (Manon et al., 1998) allow the passage of solutes with a MW of 1 to 1.5 kDa (Jung et al., 1997; Zoratti and Szabó, 1995). However, while the PTP is opened by PO₄ and Ca²⁺ and closed by cyclosporine (Bernardi et al., 1994; Bernardi and Petronilli, 1996), the YMUC is closed by PO₄ and is not affected by Ca²⁺ or cyclosporine (Jung et al., 1997). In yeast mitochondria, ATP induces the opening of the YMUC and thus, the collapse of the $\Delta\Psi$ when K⁺ is present and PO4 is absent, suggesting that yeast mitochondria accumulate K⁺ when the YMUC is open (Manon and Guérin, 1998).

It was decided to evaluate the correlation between osmotic swelling and $^{86}\text{Rb}^+$ uptake in order to better define the K⁺ uptake mechanisms in yeast mitochondria. As described in the literature (Manon *et al.*, 1998), addition of different concentrations of PO₄ resulted in the closure of the YMUC. In contrast, protein concentrations in the range used in our experiments exhibited only mild effects on the permeability status on mitochondria. Under the conditions where the YMUC remained closed, a K⁺ uptake system was unmasked, which probably is the equivalent to the

mammalian K⁺ uniporter and which is Mg²⁺ and quininesensitive. The uptake of ⁸⁶Rb⁺ detected under these conditions did not lead to modification in the light scattering properties of the mitochondrial suspension.

MATERIALS AND METHODS

Materials

All chemicals were of the best quality commercially available. Mannitol, MES, KCl, RbCl, MgCl₂, Quinine, Mersalyl and FCCP, were from Sigma Chemicals (St. Louis, MO). ⁸⁶RbCl was acquired from NEN Life Science Products, Inc. (Boston, MA). The yeast *Saccharomyces cerevisiae* was a commercial baker's strain (La Azteca, S.A).

Isolation of Yeast Mitochondria

Yeast cells were incubated in a rich liquid medium (De Kloett et al., 1961) under aeration (3 L/min) for 8 h and starved overnight in distilled water under aeration. After incubation, cells were harvested and suspended in cold, isotonic, isolation medium containing 0.6 M mannitol, 0.1% bovine serum albumin, 5 mM MES, pH 6.8 (TEA). The suspension was mixed with an equal volume of glass beads (0.45 mm diameter) and disrupted in a Braun cell homogenizer (Melsungen, Germany) and mitochondria were isolated from the homogenate by differential centrifugation as described before (Peña et al., 1977). The concentration of mitochondrial protein was determined by biuret (Gornal et al., 1949). In the experiments reported here, 2 μ L/mL of 10% oxygen peroxide were added in order to ensure the availability of oxygen throughout the experiment.

Mitochondrial Swelling

The variations in mitochondrial volume were followed by measuring the decrease in optical density at a wavelength of 540 nm in an Aminco DW2000 spectrophotometer in split mode. The spectrophotometer was equipped with a magnetic stirrer.

Transmembrane Potential

Safranine-O was used to measure the transmembrane potential following the change in absorbance at 511–533 nm in an DW2000 Aminco spectrophotometer in dual mode (Akerman and Wikström, 1976).

⁸⁶Rb⁺ Transport

The uptake of rubidium was measured after preincubating mitochondria for 1 min in 0.6 M mannitol, 5 mM MES, pH 6.8 (TEA) plus the indicated concentrations of PO₄, pH 6.8 (TEA), MgCl₂, quinine and ZnCl₂. Then, 86RbCl was added to the final concentration indicated in each figure. After a second minute of incubation, 20 mM ethanol was added and 100 μ L aliquots were taken at the indicated times. These samples were transferred to a nitrocellulose filter (0.45 mm pore) previously humidified and mounted on a high-vacuum Millipore multifilter. The filters were then washed using a buffer containing 0.6 M mannitol, 5 mM MES, 80 mM KCI, pH 6.8 (TEA) and transferred to scintillation vials, then 5 mL of scintillation liquid were added and radioactivity was measured in a Beckman LS6500 scintillation counter using the phosphate window.

RESULTS

Mitochondrial Permeability Regulation by Inorganic Phosphate (PO₄)

In the absence of inorganic phosphate, yeast mitochondria swell in the presence of K⁺ (Velours *et al.*, 1977). Under the conditions employed in this study, swelling was reevaluated, as a high concentration of mitochondria (3 mg/mL) was used in all experiments; in agreement with the literature, in the absence of PO₄ and in the presence of 20 mM KCI (Results not shown) or RbCl (Fig. 1) optical density decreased, indicating mitochondrial swelling because of ion fluxes through an open unspecific pore (Castrejón *et al.*, 1997; Manon *et al.*, 1998). This effect was energy-dependent, as indicated by the lack of swelling in the presence of the uncoupling agent FCCP (Fig. 1).

In order to further evaluate a possible effect of protein concentration on the opening state of the YMUC, the transmembrane potential was measured in the presence of different PO₄ concentrations and increasing protein (Fig. 2). In the absence of PO₄, the transmembrane potential was depleted within 1 min at all the mitochondrial concentrations tested, although an initial increase in the potential was observed which was higher as protein increased from 0.5 to 3 mg protein/mL (Fig. 2(A)). When the same experiment was performed at 0.4 mM PO₄ increasing protein resulted in a partial stabilization of the transmembrane potential. From 0.5 to 2.0 mg protein/mL, the potential still decreased to minimal values after 40 s, while at 3.0 mg protein/mL, depletion of the potential was

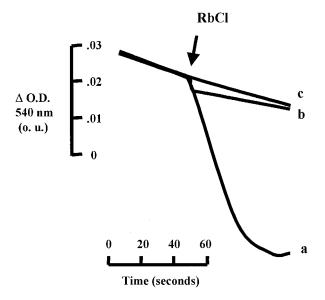
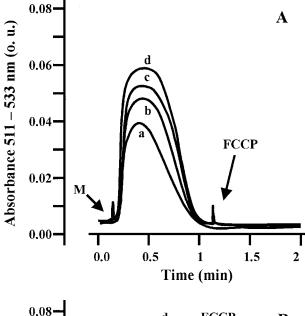


Fig. 1. Rubidium-mediated swelling of yeast mitochondria – Reaction mixture: 0.6 M mannitol, 0.5 mM MES; pH 6.8 (TEA), 20 mM ethanol, 2 μ L/mL H₂O₂ (10%), mitochondria 3 mg protein/mL. Where indicated concentrations used were (a) 20 mM RbCl, (b) 20 mM RbCl plus 6 μ M FCCP, and (c) 0 RbCl. The arrow indicates rubidium addition. Swelling was determined spectrophotometrically at 550 nm, using an Aminco DW2000 spectrophotometer in split mode. Final volume was 2 mL. (Experiment was carried out at room temperature).

only partial (Fig. 2(B)). Transmembrane potential depletion was not due to lack of oxygen, because oxygen peroxide was added in all experiments in sufficient quantity to last throughout the experiment. This was evaluated by oxymetry using a Clark electrode (Results not shown). At 4 mM PO₄, the transmembrane potential was stable at all the protein concentrations tested (Results not shown).

When increasing concentrations of PO₄ were added to the reaction mixture, the extent of mitochondrial swelling in response to the addition of 20 mM RbCl (Fig. 3) or KCl (Results not shown) decreased proportionally; swelling was almost completely inhibited at 1.0 mM PO₄. In this regard, different groups have reported that at 0.4 mM PO₄, the YMUC remains open most of the time, while at 4 mM PO₄, the YMUC is closed (Manon and Guérin, 1997).

To further characterize the effect of PO₄ on K⁺ uptake by yeast mitochondria, the uptake of ⁸⁶Rb⁺ was evaluated in the absence of PO₄, in the presence of 0.4 mM PO₄, which is known to allow for K⁺ mediated swelling, and at 4 mM PO₄, where no swelling is observed. At 0.4 mM PO₄ the YMUC is mostly opened, while at 4 mM PO₄ the YMUC seems to be closed (Jung *et al.*, 1997; Velours *et al.*, 1977). In the absence of PO₄ no uptake of ⁸⁶Rb⁺ could be detected (Fig. 4, trace a). In contrast, in the presence of 0.4 mM PO₄, mitochondria accumulated



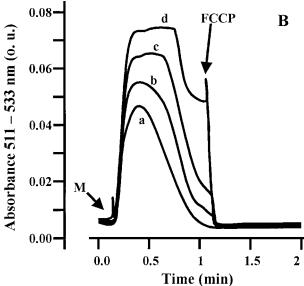


Fig. 2. Effect of protein concentration on the mitochondrial transmembrane potential – Experimental conditions were as in Fig. 1, except for 10 μ M safranine-O and different concentrations of mitochondrial protein as follows: (a) 0.5 mg/mL, (b) 1 mg/mL, (c) 2 mg/mL, (d) 3 mg/mL. Final volume was 2 mL. Mitochondria (M) or 6 μ M FCCP were added where indicated. (A) 0 and (B) 0.4 mM PO₄. PO₄ was added from either a 0.1 M or a 1 M stock solution, pH 6.8 (TEA).

approximately 10 nmoles of ⁸⁶Rb⁺ (mg protein)⁻¹ in 2 min (Fig. 4, trace b). At 4 mM PO₄, the initial rate of uptake was similar to that observed at 0.4 mM PO₄; however, at 4 mM PO₄, ⁸⁶Rb⁺ uptake slowed down after 30 s such that at 30 s, 5 nmoles ⁸⁶Rb⁺ (mg protein)⁻¹ were accumulated by mitochondria and only a small increase in uptake was observed after 2 min of incubation (Fig. 4, trace

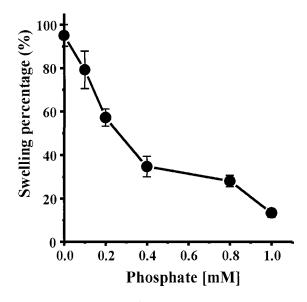


Fig. 3. Effect of PO_4 on the Rb^+ -mediated swelling of yeast mitochondria – Experimental conditions were as in Fig. 1, except that RbCl was 20 mM and PO_4 concentrations were as indicated. PO_4 was added from either a 0.1 M solution or a 1 M stock solution, pH 6.8 (TEA). Data are expressed as percentages of the swelling in the absence of PO_4 . Data are the means of four determinations \pm standard deviation.

c). In all cases, the addition of FCCP inhibited the uptake of $^{86}\text{Rb}^+$ (Results not shown). Taken together, the results indicate that PO₄ inhibits the K⁺ mediated swelling of mitochondria, while in contrast it promotes the uptake of the

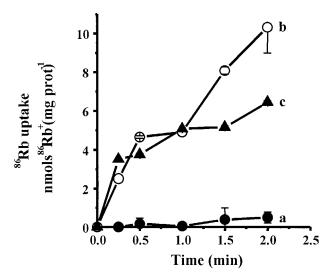


Fig. 4. Effect of PO₄ on the uptake of ⁸⁶Rb⁺ by yeast mitochondria–Experimental conditions were as in Fig. 1, except for the addition of 20 mM ⁸⁶RbCl, 2 μ L/mL H₂O₂ (10%). Final volume was 1 mL. PO₄ concentrations were (a) 0, (b) 0.4 mM, and (c) 4 mM. PO₄ was added from either a 0.1 M solution or a 1 M stock solution, pH 6.8 (TEA). Data are the means of six determinations \pm standard deviation.

radioactive K⁺ analog ⁸⁶Rb⁺ i.e, there was no correlation between both parameters. In addition, the ⁸⁶Rb⁺ uptake experiments indicated that there is an energy-dependent system for K⁺ uptake, which may be studied in the presence of the PO₄ concentrations known to close the YMUC.

Regulation of Mitochondrial Permeability by Mg²⁺ and Quinine

In mammalian mitochondria, Mg²⁺ and quinine have been reported to inhibit the uptake of K⁺. In an effort to further characterize the K⁺ uptake system in yeast mitochondria, the effects of Mg²⁺ and quinine on the Rb⁺-mediated swelling of yeast mitochondria were tested in the presence of 0.4 mM PO₄ where an active K⁺-or Rb⁺-mediated swelling has been reported. It was observed that swelling was inhibited by increasing concentrations of Mg²⁺, reaching 50% inhibition at 1 mM Mg²⁺ and stabilizing at a 70% inhibition at 2–8 mM Mg²⁺ (Fig. 5(A)). When mitochondrial swelling was determined in the presence of increasing quinine concentrations, swelling decreased linearly with quinine concentrations, reaching 55% inhibition at 0.5 mM quinine (Fig. 5(B)).

In order to assess the correlation between the swelling experiments and monovalent cation uptake, the uptake of ⁸⁶Rb⁺ was measured at 0.4 and 4 mM PO₄ and in the presence of increasing concentrations of Mg²⁺ or quinine (Fig. 6). In the presence of increasing concentrations of Mg²⁺, the uptake of ⁸⁶Rb⁺ decreased both in the presence of 0.4 mM PO₄ (Fig. 6(A), trace a) and 4 mM PO₄ (Fig. 6(A), trace b). The maximal inhibition was reached at both PO₄ concentrations at 2 mM Mg²⁺, although the Mg²⁺ mediated inhibition of uptake was higher at 4 mM PO₄, reaching 1.62 nmoles ⁸⁶Rb⁺ (mg protein)⁻¹, than at 0.4 mM PO₄ where the uptake was 4.01 nmoles ⁸⁶Rb⁺ (mg protein)⁻¹. Quinine also inhibited the uptake of ⁸⁶Rb⁺, although, as expected from the swelling experiments, the effects were lower than with Mg²⁺. In addition, the quinine-mediated inhibition of ⁸⁶Rb⁺ uptake was higher in the presence of 4 mM PO₄ (Fig. 6(B), trace b) than in the presence of 0.4 mM PO₄ (Fig. 6(B), trace a). The maximal inhibition of 86Rb+ uptake was reached in the presence of 0.5 mM quinine and 4 mM PO₄, where uptake was 2.3 nmoles ⁸⁶Rb⁺ (mg protein)⁻¹. Thus, in the experiments using increasing concentrations of different Mg²⁺ or quinine, the results obtained from the mitochondrial swelling and the 86Rb+ uptake experiments did correlate, indicating that both molecules inhibited the uptake of monovalent cations by yeast mitochondria. In this regard, the K⁺ uptake system from yeast mitochondria would be similar to its mammalian counterpart.

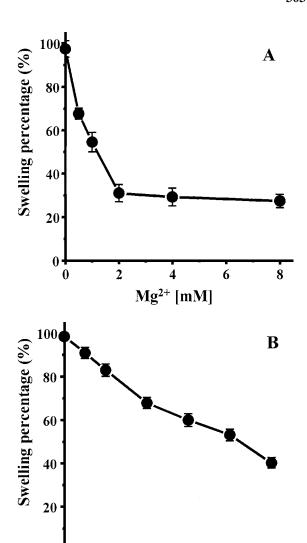


Fig. 5. Effect of $\mathrm{Mg^{2+}}$ and Quinine on the Rb⁺-mediated swelling of yeast mitochondria – Reaction mixture used was the same as in Fig. 1, except that PO₄ concentration was 0.4 mM and either (A) different magnesium concentrations were added as indicated from a 1M MgCl₂ stock solution or (B) different quinine concentrations were added from a 50 mM stock solution in dimethyl formamide. Data are expressed as percentages of maximum swelling. Data are the means of four determinations \pm standard deviation.

0.2

0.0

0.1

0.3

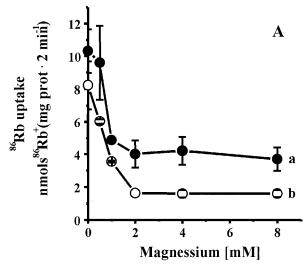
Quinine [mM]

0.4

0.5

Effect of Zn²⁺ on the Uptake of ⁸⁶Rb⁺ by Yeast Mitochondria

The rate of ⁸⁶Rb⁺ uptake by yeast mitochondria was far from the equilibrium predicted by the Nerst equation. This may be due to the slow rate of transport and/or to the establishment of an equilibrium where the



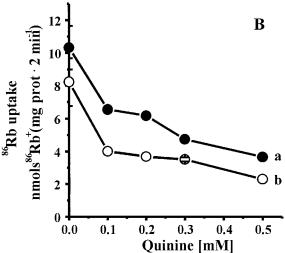


Fig. 6. Effect of Mg^{2+} or Quinine on the uptake of $^{86}Rb^+$ by yeast mitochondria – Experimental conditions were as in Fig. 3 except that (A) different magnesium concentrations were added as indicated from a 1M $MgCl_2$ stock solution and (B) different quinine concentrations were added from a 50 mM stock solution in dimethyl formamide. PO₄ concentrations were (a) 0.4 mM or (b) 4 mM. Data are expressed as the uptake of $^{86}Rb^+$ after 2 min. Data are the means of six determinations \pm standard deviation.

antiporter is actively expelling ⁸⁶Rb⁺. To explore this possibility, the uptake of ⁸⁶Rb⁺ was measured in the presence of 0.5 mM Zn²⁺, which inhibits the antiporter in yeast mitochondria (Manon and Guérin, 1992, 1995) (Fig. 7). In the presence of 4 mM PO₄, the addition of 0.5 mM Zn²⁺ resulted in an increase in the total amount of ⁸⁶Rb⁺ uptake. Thus, the inhibition of the antiporter resulted in displacement of the equilibrium of ⁸⁶Rb⁺ movements in mitochondria allowing for enhanced uptake and suggesting that this monovalent

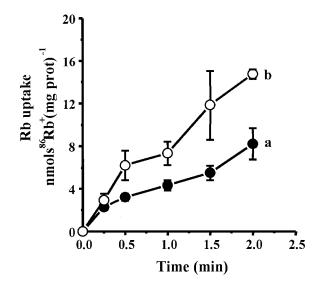


Fig. 7. 86 Rb⁺ transport by yeast mitochondria—*Effect of Zn*⁺⁺ – Experimental conditions were as in Fig. 4. Except that PO₄ concentration was 4 mM. ZnCl₂ was added from a 0.1 mM stock solution to a final concentration of (a) 0 or (b) 0.5 mM. Data are the means of six determinations \pm standard deviation.

cation uptake system is active under our experimental conditions.

DISCUSSION

Two different K+ uptake systems were detected in yeast mitochondria. These were the yeast mitochondrial unespecific channel (YMUC) and a second pathway, which was detectable under conditions where the YMUC was closed. The second pathway is probably a uniporter for K⁺ and Rb⁺, as the uptake of ⁸⁶Rb⁺ was inhibited after 30 s under the conditions where the YMUC was closed and counter-ion movements were inhibited (Fig. 4, trace c). Another evidence favoring an electrogenic uniport mechanism was the dependence of ⁸⁶Rb⁺ transport on the $\Delta\Psi$, i.e. when $\Delta\Psi$ was depleted in the presence of an uncoupler or because of the absence of PO₄, no uptake of ⁸⁶Rb⁺ was observed. The swelling data did not always correlate with the ⁸⁶Rb⁺ uptake results, i.e. at 4 mM PO₄, no swelling was detected while the uptake of ⁸⁶Rb⁺ proceeded for the first 30 s and then reached a plateau. The lack of swelling probably results from the specificity of the ⁸⁶Rb⁺ uptake system which does not allow for counterion movements and from its inhibition after 30 s; i.e. both the PO₄-mediated lack of swelling and ⁸⁶Rb⁺ uptake inhibition may be due to the impermeability of the membrane to chloride once the YMUC is closed by 4 mM PO₄.

Both Mg^{2+} and quinine inhibited the uptake of monovalent cations in yeast mitochondria. Mg^{2+} was a better

inhibitor than quinine. The ⁸⁶Rb⁺ uptake sensitivity to both Mg²⁺ and quinine increased when the YMUC was closed, while it was enhanced by Zn²⁺, suggesting that the slow ⁸⁶Rb⁺ uptake was probably counteracted by ⁸⁶Rb⁺ efflux through the antiporter (Villalobo *et al.*, 1981). It was not possible to follow the uptake of ⁸⁶Rb⁺ for longer than 2 min because of the possibility of O₂ depletion, even in the presence of added hydrogen peroxide. The highest ⁸⁶Rb⁺ uptake was observed at the low PO₄ concentration (0.4 mM), where the YMUC seems to be mostly opened, but does not seem to be as unspecific as in the absence of added PO₄. The Mg²⁺ and quinine sensitivity suggests that the K⁺ uptake system in yeast mitochondria is qualitatively similar to the mammalian uniporter (Beavis *et al.*, 1993).

In mammalian mitochondria, PTP opening results in $\Delta\Psi$ depletion and Ca^{2+} efflux; it has been suggested that Ca^{2+} detoxyfication may be a physiological function of the PTP in cells that are not programmed to die (Crompton and Costi, 1990). In yeast mitochondria, the opening of the YMUC also results in depletion of the $\Delta\Psi$ and efflux of K^+ and probably other cations. These results suggest a general physiological role for unspecific mitochondrial pores, which may be involved in cation detoxification (Crompton and Costi, 1990). Furthermore, whether yeast undergo a process resembling apoptosis or not, is still controversial and ion detoxification would be a likely alternative as a putative function of mitochondrial unspecific channels (Manon *et al.*, 1998).

In the absence of PO₄, where the YMUC is open and the transmembrane potential is depleted, K⁺-mediated swelling has been observed, (Castrejón *et al.*, 1997; Manon and Guérin, 1997). In the absence of PO₄, no ⁸⁶Rb⁺ accumulation was observed, even though swelling was present. Under these conditions, the YMUC was completely open, probably allowing for rapid permeation of molecules, leaving only proteins inside the matrix and evoking swelling as a result of oncotic pressure. Another possibility, is that ⁸⁶Rb⁺ was taken up leading to swelling, but the filtration process involving washing filters with cold Rb⁺ resulted in the exit of ⁸⁶Rb⁺ through the open YMUC.

In the presence of 0.4 mM PO₄, the transmembrane potential was lost partially only if K⁺ was added to the reaction mixture, suggesting that the unspecific pore became rather selective at this PO₄ concentration. Under this condition, both swelling and ⁸⁶Rb⁺ uptake correlated. At 4 mM PO₄, the data were still different, as no swelling was detected, while the uptake of ⁸⁶Rb⁺ was only slightly less than at 0.4 mM PO₄. At 4 mM PO₄, the unspecific channel is closed and thus the uptake of K⁺ seemed to be catalyzed by a different system, which was less active

than the YMUC. At 4 mM PO₄, ⁸⁶Rb⁺ did not promote swelling, probably due to the exit of K⁺ through the reported K⁺/H⁺ antiporter (Jung *et al.*, 1997), which would be active in view of the high transmembrane potential detected at this PO₄ concentration.

Thus, in yeast, the K⁺ recycling pathways seem to resemble mammalian mitochondria, in that they keep matrix K⁺ at levels similar, or lower than those found in the cytosol (Brierley *et al.*, 1994; Garlid, 1996). In mammalian mitochondria, K⁺ enters the matrix through a uniport that opens when ATP is depleted and is energized by the transmembrane potential (Beavis *et al.*, 1993). Once in the matrix, K⁺ is expelled to the cytoplasm throughout a K⁺/H⁺ uniport (Brierley *et al.*, 1984, 1994). In yeast, the K⁺/H⁺ antiporter has already been described (Jung *et al.*, 1997; Villalobo *et al.*, 1981), while the uptake system described here would serve an equivalent function to that of the mammalian uniporter. When the YMUC was closed, the uptake of ⁸⁶Rb⁺ increased its sensitivity to the mammalian K⁺ uptake inhibitors, Mg²⁺ and quinine (Fig. 6).

Under our experimental conditions, the rate of ⁸⁶Rb⁺ uptake was slow, and thus it was not possible to reach equilibrium before the transmembrane potential collapsed, once oxygen was depleted, even in the presence of oxygen peroxide. Nonetheless, addition of the antiporter inhibitor Zn²⁺ (Manon and Guérin, 1992, 1995) led to higher uptake, suggesting that the uptake of ⁸⁶Rb⁺ was counter balanced by an active exit of ⁸⁶Rb⁺ through the uniporter.

A ⁸⁶Rb⁺ uptake system was observed in yeast mitochondria under conditions where the unspecific pore is closed. This may be the system used by mitochondria to take up K⁺ when in the presence of a high membrane potential. This transporter resembles the mammalian uniporter in its energy dependence as well as in the sensitivity to Mg²⁺ and to quinine.

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REFERENCES

Akerman, K. E. O., and Wikström, K. F. (1976). FEBS Lett. 68, 191–197.
 Ballarin, C., and Sorgato, M. C. (1995). J. Biol. Chem. 270, 19262–19268.

Ballarin, C., and Sorgato, M. C. (1996). J. Bioenerg. Biomembr. 28, 125–130.

Beavis, A. D., Brannan, R. D., and Garlid, K. D. (1993). *J. Biol. Chem.* **260**, 13424–13433.

- Beavis, A. D., Lu, Y., and Garlid, K. D. (1993). J. Biol. Chem. 68, 997– 1004.
- Bernardi, P., Broekemeier, K. M., and Pfeiffer, D. R. (1994). *J. Bioenerg. Biomembr.* **26**, 509–517.
- Bernardi, P., and Petronilli, V. (1996). *J. Bioenerg. Biomembr.* 28, 131–138.
- Brierley, G. P. (1978). In *The Molecular Biology of Membranes* (Fleischer, S., Hatefi, Y., McLennan, D. H., and Tzagaloff, A., eds.), Plenum, New York, pp. 295–308.
- Brierley, G. P., Baysal, K., and Jung, D. W. (1994). J. Bioenerg. Biomembr. 26, 519–526.
- Brierley, G. P., and Jung, D. W. (1988). *J. Bioenerg. Biomembr.* **20**, 193–208.
- Brierley, G. P., Jurkowitz, M. S., Farooqui, T., and Jung, D. W. (1984).
 J. Biol. Chem. 259, 14672–14678.
- Castrejón, V., Parra, C., Moreno, R., Peña, A., and Uribe, S. (1997). *Arch. Biochem. Biophys.* **346**, 37–44.
- Chappel, J. B., and Crofts, A. R. (1966). In Regulation of Metabolic Process in Mitochondria, Vol. 7 (Tager, J. M., Papa, S., Quagliarello, E., and Slater, E. C., eds.), Elsevier, Amsterdam, pp. 293–312.
- Crompton, M., and Costi, A. (1990). Biochem. J. 266, 33-39.
- De Kloett, S. R., Van Wermeskerken, R. K. A., and Konigsberg, V. V. (1961). *Biochim. Biophys. Acta.* 47, 138–143.
- Diwan, J. J. (1986). Biochem. Biophys. Res. Comm. 135, 830-836.
- Diwan, J. J., Haley, T., and Sanadi, D. R. (1988). Biochem. Biophys. Res. Comm. 153, 224–230.
- Garlid, K. D. (1980). J. Biol. Chem. 265, 11273-11279.
- Garlid, K. D. (1988). In *Integration of Mitochondrial Function* (Lemasters, J. J., Hackenbroock, C. R., Thruman, R. G., and Westerhoff, H. V., eds.) Plenum, New York, pp. 259–268.
- Garlid, K. D. (1996). Biochim. Biophys. Acta 1275, 123-126.
- Gornal, G. A., Bardaville, J. C., and David, M. M. (1949). J. Biol. Chem. 177, 752–766.
- Halestrap, A. P. (1989). Biochim. Biophys. Acta 973, 355-382.

- Inoue, I., Nagase, H., Kishi, K., and Higuti, T. (1991). *Nature (London)* 352, 244–247.
- Jezek, P., Mahdi, F., and Garlid, K. D. (1990). J. Biol. Chem. 25, 10522– 10536
- Jung, D. W., Bradshaw, P. C., and Pfeiffer, D. R. (1997). J. Biol. Chem. 272, 21104–21112.
- Jung, D. W., and Brierley, G. P. (1986). J. Biol. Chem. 261, 6408–6415.Kapus, A., Szaszi, K., Kaldi, K., Ligeti, E., and Fonyo, A. (1990). J. Biol. Chem. 265, 18063–18066.
- Manon, S., and Guérin, M. (1992). Biochim. Biophys. Acta 1108, 169– 176.
- Manon, S., and Guérin, M. (1995). Biochem. Mol. Biol. Int. 35, 585–593.
 Manon, S., and Guérin, M. (1997). Biochim. Biophys. Acta 1318, 317–321.
- Manon, S., and Guérin, M. (1998). *Biochem. Mol. Biol. Int.* 44, 565–575.
 Manon, S., Roucou, X., Guérin, M., Rigoulet, M., and Guérin, B. (1998).
 J. Bioenerg. Biomembr. 30, 419–429.
- Martin, W. H., DiResta, D. J., and Garlid, K. D. (1986). J. Biol. Chem. 261, 12300–12305.
- Mitchell, P. (1961). Nature 191, 144-148.
- Nakashima, R. A., and Garlid, K. D. (1982). J. Biol. Chem. 257, 9252–9254.
- Nicholls, D. G., Grav, H. J., and Lindberg, D. (1972). *Eur. J. Biochem.* **31**, 526–533.
- Peña, A., Piña, M. Z., Escamilla, E., and Piña, E. (1977). *FEBS Lett.* **80**, 209–213.
- Roucou, X., Manon, S., and Guérin, M. (1995). *J. Bioenerg. Biomem.* **27**, 353–362.
- Velours, J., Rigoulet, M., and Guérin, B. (1977). FEBS Lett. 81, 18–22.Villalobo, A., Briquet, M., and Goffeau, A. (1981). Biochim. Biophys. Acta 637, 124–129.
- Welihinda, A. A., Trumbly, R. J., Garlid, K. D., and Beavis, A. D. (1993). *Biochim. Biophys. Acta* 1144, 367–373.
- Zoratti, M., and Szabó, I. (1995). Biochim. Biophys. Acta 1241, 139-176.